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most likely and
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Einstein solids 2-29

Use the Sackur-Tetrode equation to calculate the

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2. As in the Otto cycle, the red and blue curves are

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adiabats, so no heat is exchanged during these steps. The efficiency of the Diesel cycle is most easily worked out by calculating the heat input Q_1 along the yellow step 2 to 3 and the heat expelled during the green step 4 to 1. The efficiency is then

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$e = W - Q$

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DIESEL ENGINES

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(in Schroeder), DH

$H_{2O} = 285.83 \text{ kJ}$

mol^{-1} and $DH =$

$890.36 \text{ kJ mol}^{-1}$

(12) This time, the

final volume is $\frac{1}{3}$

of the initial

volume, since the 2

moles of water has

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condensed out as liquid with negligible volume compared to the gases. Thus the atmosphere does work $P\Delta V = RT\Delta n$
(13) = 8.31 J K⁻¹
(298 K)(2) (14) =
4.953 kJ (15)

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