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Questions 1 through 5
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Answers for the
questions in Chapter 8:
ROUND 3 Exercise 8-1:
Using the activity series,
predict and balance the
following single
replacement reactions.

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Use abbreviations to indicate the appropriate phase of reactants and products where possible.

Note: Not all of the 11 reactions will occur. For those that do not, write no reaction. 1.

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the questions in Chapter 8: ROUND 3 Exercise 8-1: Using the activity series, predict and balance the following single replacement reactions. Use abbreviations to indicate the appropriate phase of reactants and products where possible.

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Answers Chapter 10 ...

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Determine the oxidation
number of the

Ch 11
underlined element:

KMnO_4 Since K is an
alkali metal, its charge
must be $1+$ Oxygen is $2-$
but there are four of
them, therefore, 4 times
 $2-$ equals $8-$ If $1+$ and $8-$
are added

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Since many of the
transuranium elements
were prepared at the
Lawrence Radiation
Center, University of
California at Berkeley,
California, U.S.A., it is

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logical to find the following names:
Americium—named after the Americas (after all, one must be patriotic!).

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Precipitation

Reactions(See solubility
rules) - Pg. 28, Practice-

Pg. 49 □ Formation of a

Gas - Pg. 50, Practice-

Pg. 51 □ Acid Base

Neutralization

Reactions- Pg. 53,

Practice-Pg.54 oWeak

acids and bases do not

break apart and are in

net ionic equation. □

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Synthesis and

Decomposition

Reactions - Pgs. 42-43,

Practice- Pg. 43 □ Single

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Edition. Authors: 11

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Jane D. Smith,

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new and revised,

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offering thorough,
comprehensive
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that provide continuous
reinforcement to
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